

Table 2. Selected geometric parameters (Å, °)

|           |            |             |           |
|-----------|------------|-------------|-----------|
| S—C2      | 1.742 (3)  | C8—N9       | 1.375 (4) |
| S—C11     | 1.821 (3)  | N9—C1'      | 1.456 (4) |
| N1—C2     | 1.364 (4)  | C10—C11     | 1.496 (5) |
| N1—C6     | 1.396 (4)  | C1'—O4'     | 1.407 (3) |
| N1—C10    | 1.473 (4)  | C1'—C2'     | 1.531 (3) |
| C2—N3     | 1.311 (4)  | C2'—O2'     | 1.406 (3) |
| N3—C4     | 1.360 (4)  | C2'—C3'     | 1.522 (4) |
| C4—N9     | 1.359 (4)  | C3'—O3'     | 1.424 (3) |
| C4—C5     | 1.386 (4)  | C3'—C4'     | 1.537 (4) |
| C5—N7     | 1.382 (4)  | C4'—O4'     | 1.446 (4) |
| C5—C6     | 1.424 (4)  | C4'—C5'     | 1.508 (4) |
| C6—O6     | 1.233 (4)  | C5'—O5'     | 1.410 (4) |
| N7—C8     | 1.308 (4)  |             |           |
| C2—S—C11  | 91.36 (15) | C4—N9—C8    | 106.2 (2) |
| C2—N1—C6  | 124.5 (2)  | C4—N9—C1'   | 128.8 (2) |
| C2—N1—C10 | 114.6 (2)  | C8—N9—C1'   | 125.0 (2) |
| C6—N1—C10 | 120.7 (2)  | N1—C10—C11  | 107.7 (3) |
| N3—C2—N1  | 125.6 (2)  | C10—C11—S   | 106.7 (2) |
| N3—C2—S   | 121.6 (2)  | O4'—C1'—N9  | 108.2 (2) |
| N1—C2—S   | 112.8 (2)  | O4'—C1'—C2' | 105.3 (2) |
| C2—N3—C4  | 111.8 (2)  | N9—C1'—C2'  | 114.9 (2) |
| N9—C4—N3  | 127.0 (3)  | O2'—C2'—C3' | 115.7 (2) |
| N9—C4—C5  | 105.8 (2)  | O2'—C2'—C1' | 114.8 (2) |
| N3—C4—C5  | 127.2 (3)  | C3'—C2'—C1' | 101.3 (2) |
| N7—C5—C4  | 110.6 (2)  | O3'—C3'—C2' | 110.9 (2) |
| N7—C5—C6  | 129.7 (3)  | O3'—C3'—C4' | 109.0 (2) |
| C4—C5—C6  | 119.7 (2)  | C2'—C3'—C4' | 103.1 (2) |
| O6—C6—N1  | 119.8 (3)  | O4'—C4'—C5' | 107.6 (2) |
| O6—C6—C5  | 129.0 (3)  | O4'—C4'—C3' | 106.4 (2) |
| N1—C6—C5  | 111.1 (3)  | C5'—C4'—C3' | 115.5 (3) |
| C8—N7—C5  | 104.0 (2)  | C1'—O4'—C4' | 109.6 (2) |
| N7—C8—N9  | 113.3 (3)  | O5'—C5'—C4' | 113.1 (3) |

Table 3. Contact distances (Å)

| A                 | H    | D   | A...H    | D...A     |
|-------------------|------|-----|----------|-----------|
| N3                | H5O' | O5' | 2.21 (6) | 3.045 (4) |
| O <sup>j</sup>    | H3O' | O3' | 1.93 (4) | 2.725 (4) |
| O6                | H1   | O   | 2.03 (4) | 2.784 (3) |
| O3' <sup>ii</sup> | H2   | O3  | 1.98 (6) | 2.826 (4) |
| O5' <sup>i</sup>  | H2O' | O2' | 1.98 (4) | 2.742 (3) |

Symmetry codes: (i)  $x, y, 1 - z$ ; (ii)  $1 - x, y, 1 + z$ .

Dispersion corrections and absorption coefficients were taken from *International Tables for Crystallography* (1992, Vol. C, Tables 6.1.1.4, 4.2.6.8 and 4.2.4.2). Since (I) crystallizes in a polar space group, polar-axis restraints were applied by the method of Flack & Schwarzenbach (1988) and the absolute structure of the crystal was established as described by Flack (1983), which is consistent with the known absolute configuration of  $\beta$ -D-ribofuranose. Data collection and cell refinement: Rigaku AFC-5 software. Data reduction: *NRCVAX* (Gabe, Lee & Le Page, 1985). Program used to solve structure: *SHELXS86* (Sheldrick, 1990). Program used to refine structure: *SHELXL* (Sheldrick, 1994). Molecular graphics: *ORTEP* (Johnson, 1971). Software used to prepare material for publication: *SHELXL*; *NRCVAX*.

This work was supported by the National Science Council of the Republic of China (NSC81-0203-B001-14 to the Institute of Molecular Biology and NSC79-042-B016-136 to J-WC).

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71785 (8 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: OH1058]

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*Acta Cryst.* (1994). **C50**, 736-738

### Absolute Stereostructure of 13,17-Epoxy-*alisol B* 23-Acetate Isolated from *Alisma orientale*

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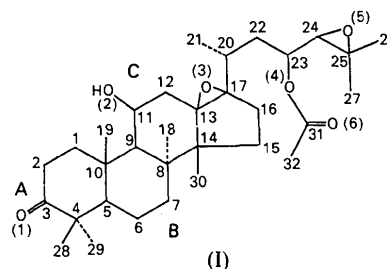
(Received 10 August 1993; accepted 2 November 1993)

## Abstract

The absolute configuration of the epoxy group in 13 $\beta$ ,17 $\beta$ :-24(*R*),25-diepoxy-1 $\beta$ -hydroxydammar-3-one 23(*S*)-acetate, C<sub>32</sub>H<sub>50</sub>O<sub>6</sub>, m.p. 472-474 K, [ $\alpha$ ]<sub>D</sub>+139.4° ( $c = 0.96$ , CHCl<sub>3</sub>), isolated from *Alisma orientale* rhizomes, was established to be the  $\beta$  orientation by X-ray crystallographic analysis. The *A*, *B* and *C* rings have chair forms. An intermolecular hydrogen bond is observed between O2 and O6' with a distance of 3.148 (6) Å.

## Comment

13,17-Epoxyalisol B 23-acetate (I) was isolated from *Alisma orientale* rhizomes along with 20 related triterpenes of the protostane type (Nakajima, Mikoshiba, Ida & Shoji, 1984; Ida, Satoh, Nakajima, Yamaguchi & Shoji, 1989).



The basic structure was established on the basis that the compound was obtained as the sole product with quantitative yield from arisol B 23-acetate (Murata, Shinohara & Miyamoto, 1970) on oxidation with *m*-chloroperbenzoic acid (Nakajima, Mikoshiba, Ida & Shoji, 1984; Ida, Satoh, Nakajima, Yamaguchi & Shoji, 1989; Fukuyama, Pei-Wu, Rei, Yamada & Nakagawa, 1988). It has been proposed that the epoxide group has the  $\beta$  orientation on the basis of the following considerations: the epoxide is formed stereospecifically as the sole product from the double bond of alisol B 23-acetate and the oxidation reaction seems to prefer the  $\beta$  side of the double bond rather than the  $\alpha$  side, since the estimated electrostatic potential is higher on the  $\beta$  side than on the  $\alpha$  side, according to our theoretical calculations (Nakajima, Mikoshiba, Ida & Shoji, 1984; Ida, Satoh, Nakajima, Yamaguchi & Shoji, 1989). The X-ray crystallographic analysis justified this proposal.

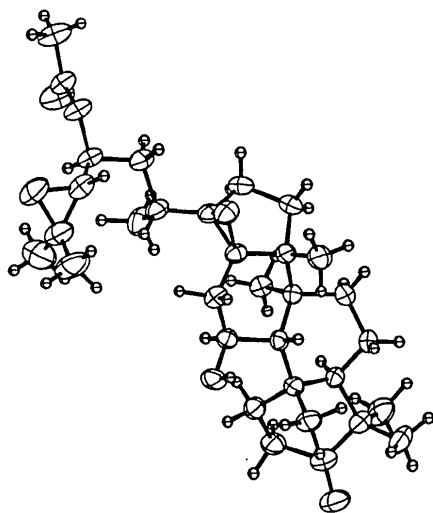


Fig. 1. Displacement-ellipsoid plot of the title molecule. Ellipsoids are drawn at the 50% probability level with isotropic H-atom parameters represented by spheres of arbitrary size.

## Experimental

The title compound was isolated from a crude sample of the Chinese drug 'Zexie'.

### Crystal data

$C_{32}H_{50}O_6$   
 $M_r = 530.75$   
 Monoclinic  
 $P2_1$   
 $a = 10.558 (1) \text{ \AA}$   
 $b = 20.351 (1) \text{ \AA}$   
 $c = 7.326 (2) \text{ \AA}$   
 $\beta = 106.22 (1)^\circ$   
 $V = 1511.3 (6) \text{ \AA}^3$   
 $Z = 2$   
 $D_x = 1.166 \text{ Mg m}^{-3}$

Cu  $K\alpha$  radiation  
 $\lambda = 1.5418 \text{ \AA}$   
 Cell parameters from 20 reflections  
 $\theta = 28.0\text{--}30.5^\circ$   
 $\mu = 0.593 \text{ mm}^{-1}$   
 $T = 297 \text{ K}$   
 Prism  
 $0.55 \times 0.35 \times 0.11 \text{ mm}$   
 Clear

### Data collection

Rigaku AFC-5 diffractometer  
 $\omega/2\theta$  scans [width (1.3 + 0.14tan $\theta$ ) $^\circ$  (in  $\omega$ ); speed 16 $^\circ \text{ min}^{-1}$ ]  
 Absorption correction: none  
 2586 measured reflections  
 2323 independent reflections  
 2149 observed reflections  
 $[F > 3\sigma(F)]$

$R_{\text{int}} = 0.018$   
 $\theta_{\text{max}} = 60^\circ$   
 $h = -11 \rightarrow 11$   
 $k = 0 \rightarrow 22$   
 $l = 0 \rightarrow 8$   
 3 standard reflections monitored every 150 reflections  
 intensity variation: < 3%

### Refinement

Refinement on  $F^2$

$R(F) = 0.040$   
 $wR(F^2) = 0.038$   
 $S = 1.98$

2149 reflections  
 488 parameters  
 All H-atom parameters refined

Calculated weights  
 $w = 1/[\sigma^2(F) + 0.007F^2]$

$(\Delta/\sigma)_{\text{max}} = 0.12$

$\Delta\rho_{\text{max}} = 0.30 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.28 \text{ e \AA}^{-3}$

Extinction correction: none  
 Atomic scattering factors from *International Tables for X-ray Crystallography* (1974, Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

$$U_{\text{eq}} = (1/3)\sum_i\sum_j U_{ij}a_i^*a_j^*a_i\cdot a_j.$$

|     | x           | y           | z           | $U_{\text{eq}}$ |
|-----|-------------|-------------|-------------|-----------------|
| C1  | 0.9375 (5)  | 0.0955      | 0.7525 (6)  | 0.045 (1)       |
| C2  | 1.0266 (5)  | 0.1524 (3)  | 0.7189 (7)  | 0.051 (1)       |
| C3  | 1.1155 (4)  | 0.1839 (3)  | 0.8955 (7)  | 0.048 (1)       |
| C4  | 1.0531 (4)  | 0.2033 (3)  | 1.0520 (6)  | 0.045 (1)       |
| C5  | 0.9239 (4)  | 0.1628 (3)  | 1.0272 (6)  | 0.036 (1)       |
| C6  | 0.8689 (4)  | 0.1648 (4)  | 1.1998 (6)  | 0.046 (1)       |
| C7  | 0.7231 (4)  | 0.1500 (3)  | 1.1346 (7)  | 0.048 (1)       |
| C8  | 0.6848 (4)  | 0.0901 (3)  | 0.9950 (5)  | 0.035 (1)       |
| C9  | 0.8104 (4)  | 0.0505 (3)  | 0.9866 (6)  | 0.033 (1)       |
| C10 | 0.9304 (4)  | 0.0915 (3)  | 0.9602 (5)  | 0.036 (1)       |
| C11 | 0.7696 (4)  | -0.0079 (3) | 0.8472 (6)  | 0.039 (1)       |
| C12 | 0.6666 (4)  | -0.0521 (3) | 0.8942 (7)  | 0.042 (1)       |
| C13 | 0.5565 (4)  | -0.0157 (3) | 0.9398 (6)  | 0.037 (1)       |
| C14 | 0.5906 (4)  | 0.0437 (3)  | 1.0690 (5)  | 0.038 (1)       |
| C15 | 0.4543 (4)  | 0.0753 (4)  | 1.0542 (7)  | 0.050 (1)       |
| C16 | 0.3553 (4)  | 0.0490 (4)  | 0.8724 (8)  | 0.050 (1)       |
| C17 | 0.4171 (4)  | -0.0133 (3) | 0.8245 (6)  | 0.040 (1)       |
| C18 | 0.6119 (4)  | 0.1166 (3)  | 0.7953 (6)  | 0.040 (1)       |
| C19 | 1.0568 (4)  | 0.0576 (4)  | 1.0799 (8)  | 0.047 (1)       |
| C20 | 0.3589 (4)  | -0.0498 (3) | 0.6411 (6)  | 0.043 (1)       |
| C21 | 0.3460 (6)  | -0.0052 (4) | 0.4678 (8)  | 0.064 (2)       |
| C22 | 0.2258 (4)  | -0.0798 (4) | 0.6428 (7)  | 0.048 (1)       |
| C23 | 0.1830 (4)  | -0.1353 (3) | 0.4982 (7)  | 0.044 (1)       |
| C24 | 0.2738 (4)  | -0.1926 (3) | 0.5534 (8)  | 0.051 (1)       |
| C25 | 0.3745 (5)  | -0.2106 (3) | 0.4574 (9)  | 0.063 (2)       |
| C26 | 0.3892 (9)  | -0.1763 (5) | 0.2857 (13) | 0.095 (3)       |
| C27 | 0.4942 (6)  | -0.2468 (5) | 0.5763 (16) | 0.089 (3)       |
| C28 | 1.1532 (6)  | 0.1977 (4)  | 1.2487 (9)  | 0.064 (2)       |
| C29 | 1.0156 (6)  | 0.2761 (4)  | 1.0154 (10) | 0.066 (2)       |
| C30 | 0.6541 (5)  | 0.0199 (4)  | 1.2750 (7)  | 0.056 (2)       |
| C31 | -0.0519 (4) | -0.1289 (3) | 0.3885 (7)  | 0.047 (1)       |
| C32 | -0.1775 (5) | -0.1547 (4) | 0.4149 (12) | 0.065 (2)       |
| O1  | 1.2284 (3)  | 0.1983 (3)  | 0.9033 (5)  | 0.072 (1)       |
| O2  | 0.8803 (2)  | -0.0488 (3) | 0.8421 (5)  | 0.056 (1)       |
| O3  | 0.4573 (3)  | -0.0566 (3) | 0.9900 (4)  | 0.051 (1)       |
| O4  | 0.0530 (2)  | -0.1588 (3) | 0.5057 (4)  | 0.049 (1)       |
| O5  | 0.2528 (3)  | -0.2487 (3) | 0.4273 (5)  | 0.068 (1)       |
| O6  | -0.0434 (3) | -0.0878 (3) | 0.2753 (5)  | 0.067 (1)       |

Table 2. Selected geometric parameters (Å, °)

|             |            |             |            |
|-------------|------------|-------------|------------|
| C1—C10      | 1.546 (6)  | C14—C15     | 1.552 (7)  |
| C1—C2       | 1.553 (8)  | C15—C16     | 1.542 (7)  |
| C2—C3       | 1.513 (7)  | C16—C17     | 1.512 (10) |
| C3—C4       | 1.526 (8)  | C17—C20     | 1.508 (7)  |
| C4—C28      | 1.535 (7)  | C20—C21     | 1.534 (9)  |
| C4—C29      | 1.537 (11) | C20—C22     | 1.536 (7)  |
| C4—C5       | 1.560 (7)  | C22—C23     | 1.529 (9)  |
| C5—C6       | 1.532 (7)  | C23—C24     | 1.490 (9)  |
| C5—C10      | 1.541 (9)  | C24—C25     | 1.477 (9)  |
| C6—C7       | 1.509 (7)  | C25—C26     | 1.483 (12) |
| C7—C8       | 1.570 (9)  | C25—C27     | 1.511 (9)  |
| C8—C18      | 1.547 (6)  | C31—C32     | 1.488 (9)  |
| C8—C9       | 1.568 (7)  | O1—C3       | 1.213 (6)  |
| C8—C14      | 1.574 (8)  | O2—C11      | 1.445 (7)  |
| C9—C11      | 1.548 (8)  | O3—C17      | 1.462 (7)  |
| C9—C10      | 1.573 (7)  | O3—C13      | 1.463 (7)  |
| C10—C19     | 1.540 (7)  | O4—C31      | 1.342 (6)  |
| C11—C12     | 1.523 (8)  | O4—C23      | 1.469 (6)  |
| C12—C13     | 1.493 (8)  | O5—C24      | 1.448 (8)  |
| C13—C17     | 1.479 (5)  | O5—C25      | 1.463 (7)  |
| C13—C14     | 1.516 (8)  | O6—C31      | 1.198 (8)  |
| C14—C30     | 1.547 (7)  |             |            |
| C10—C1—C2   | 113.1 (3)  | C17—C13—C14 | 110.9 (5)  |
| C3—C2—C1    | 116.1 (4)  | C12—C13—C14 | 118.2 (4)  |
| O1—C3—C2    | 120.7 (5)  | C13—C14—C30 | 108.9 (5)  |
| O1—C3—C4    | 121.8 (5)  | C13—C14—C15 | 103.5 (3)  |
| C2—C3—C4    | 117.2 (4)  | C13—C14—C8  | 109.0 (4)  |
| C3—C4—C28   | 111.0 (4)  | C30—C14—C15 | 109.8 (4)  |
| C3—C4—C29   | 105.3 (5)  | C30—C14—C8  | 112.3 (3)  |
| C3—C4—C5    | 108.8 (4)  | C15—C14—C8  | 113.0 (5)  |
| C28—C4—C29  | 108.1 (5)  | C16—C15—C14 | 107.9 (5)  |
| C28—C4—C5   | 114.8 (5)  | C17—C16—C15 | 105.1 (4)  |
| C29—C4—C5   | 108.3 (4)  | O3—C17—C13  | 59.7 (3)   |
| C6—C5—C10   | 110.5 (5)  | O3—C17—C20  | 113.2 (5)  |
| C6—C5—C4    | 114.2 (4)  | O3—C17—C16  | 110.9 (4)  |
| C10—C5—C4   | 114.8 (4)  | C13—C17—C20 | 125.9 (5)  |
| C7—C6—C5    | 108.7 (3)  | C13—C17—C16 | 108.6 (4)  |
| C6—C7—C8    | 114.6 (5)  | C20—C17—C16 | 121.7 (4)  |
| C18—C8—C9   | 110.1 (4)  | C17—C20—C21 | 111.5 (5)  |
| C18—C8—C7   | 108.4 (5)  | C17—C20—C22 | 109.6 (4)  |
| C18—C8—C14  | 109.8 (3)  | C21—C20—C22 | 111.8 (4)  |
| C9—C8—C7    | 111.0 (3)  | C23—C22—C20 | 112.5 (4)  |
| C9—C8—C14   | 109.1 (5)  | O4—C23—C24  | 105.8 (5)  |
| C7—C8—C14   | 108.3 (4)  | O4—C23—C22  | 108.2 (4)  |
| C11—C9—C8   | 109.8 (3)  | C24—C23—C22 | 110.3 (4)  |
| C11—C9—C10  | 114.3 (4)  | O5—C24—C25  | 60.0 (4)   |
| C8—C9—C10   | 116.8 (5)  | O5—C24—C23  | 118.1 (4)  |
| C19—C10—C5  | 110.4 (4)  | C25—C24—C23 | 124.0 (5)  |
| C19—C10—C1  | 108.2 (4)  | O5—C25—C24  | 59.0 (4)   |
| C19—C10—C9  | 107.0 (5)  | O5—C25—C26  | 114.6 (5)  |
| C5—C10—C1   | 106.5 (4)  | O5—C25—C27  | 112.8 (6)  |
| C5—C10—C9   | 110.6 (4)  | C24—C25—C26 | 123.1 (6)  |
| C1—C10—C9   | 114.2 (3)  | C24—C25—C27 | 116.4 (6)  |
| O2—C11—C12  | 107.4 (5)  | C26—C25—C27 | 116.7 (6)  |
| O2—C11—C9   | 112.5 (3)  | O6—C31—O4   | 123.4 (4)  |
| C12—C11—C9  | 112.9 (4)  | O6—C31—C32  | 125.3 (5)  |
| C13—C12—C11 | 114.0 (6)  | O4—C31—C32  | 111.3 (5)  |
| O3—C13—C17  | 59.6 (3)   | C17—O3—C13  | 60.8 (3)   |
| O3—C13—C12  | 115.5 (5)  | C31—O4—C23  | 116.2 (5)  |
| O3—C13—C14  | 111.0 (4)  | C24—O5—C25  | 61.0 (4)   |
| C17—C13—C12 | 126.9 (4)  |             |            |

Lists of structure factors, anisotropic displacement parameters and H-atom coordinates have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71803 (14 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: AS1083]

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*Acta Cryst.* (1994). **C50**, 738–740

## A Kaurane Derivative Isolated from *Alisma orientale*

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(Received 10 August 1993; accepted 2 November 1993)

## Abstract

The structure of a new diterpene, C<sub>20</sub>H<sub>30</sub>O<sub>2</sub>, m.p. 452–453 K, [ $\alpha$ ]<sub>D</sub> –28.4° (*c* = 1.0, acetone), isolated from fresh rhizomes of *Alisma orientale* (Alismataceae) was established to be 16(*R*)-(–)-kaurane-2,12-dione by means of X-ray crystallographic analysis and its optical rotation. The *A*, *B* and *C* rings have chair conformations. The puckering parameter of the bridging *D* ring, the angle between the planes C(14)—C(13)—C(16)—C(15) and C(15)—C(8)—C(14), is 39.4 (7)°.

Initial structure analysis was performed with a continuous process connected to the data collection using the fully automatic procedure *FASE* (Yamaguchi, 1993). Data collection and cell refinement were performed using *AFD* (Rigaku Corporation, 1985a); data reduction was by *FASE*. The structure was solved by direct methods, included in *FASE*, and with *SAPI85* (Yao, Zheng, Qian, Han, Gu & Fan, 1985), and refined using *RCRYSTAN* (Rigaku Corporation, 1985b). Molecular graphics were obtained using *ACV* (Stardent Computer Inc., 1990) and the material for publication was prepared with *XPACK* (Yamaguchi, 1987).